Ammonia/Ammonium Assay Kit (Colorimetric/Fluorometric)

LS-K242-100 (100 Tests) • **Store at -20°C**

LSBio
LifeSpan BioSciences, Inc.

Introduction

AMMONIA (NH_3) or its ion form ammonium (NH_4^+) is an important source of nitrogen for living systems. It is synthesized through amino acid metabolism and is toxic when present at high concentrations. In the liver, ammonia is converted to urea through the urea cycle. Elevated levels of ammonia in the blood (hyperammonemia) have been found in liver dysfunction (cirrhosis), while hypoammonemia has been associated with defects in the urea cycle enzymes (e.g. ornithine transcarbamylase).

Simple, direct and automation-ready procedures for measuring NH_3 are popular in research and drug discovery. LSBio's ammonia assay is designed to directly measure NH_3 and NH_4^+. In this assay, NADH is converted to NAD^+ in the presence of NH_3, ketoglutarate and glutamate dehydrogenase. The decrease in optical density at 340 nm or fluorescence intensity at $\lambda_{em/ex}$= 450/360 nm is directly proportionate to the NH3 concentration in the sample.

Key Features

- High sensitivity and wide linear range. Use 20 µL sample. Linear detection range 24 to 1000 µM ammonia.
- Homogeneous and simple procedure. Simple "mix-and-measure" procedure allows reliable quantitation of NH_3 within 30 minutes.

Applications

- Direct Assays: NH_3 in biological samples (e.g. serum, plasma, urine, saliva, cell culture etc).

Components

Component	K242-100
	100 Tests
Assay Buffer	20 mL
Ketoglutarate	120 µL
NADH Reagent	Dried
Enzyme	120 µL
Standard	400 µL

Materials Not Supplied

Pipetting devices, and clear flat-bottom 96-well plates and optical density plate reader for colorimetric assays; black flat-bottom 96-well plate and fluorescence intensity plate reader for fluorometric assays.

Storage

The kit is shipped on ice. Store all kit components at -20 °C.

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Assay Procedure

Reagent Preparation. Equilibrate all components to room temperature. Briefly centrifuge all tubes before opening. Reconstitute the NADH Reagent tube with 1000 µL dH_2O (final 10 mM). Unused reconstituted NADH reagent is stable for three weeks when stored frozen at -20°C.

Sample preparation: solid samples can be extracted by homogenization in distilled water (dH_2O) and filtered, centrifuged or, if necessary, deproteinized to remove any undissolved material. Samples should be clear and colorless with pH adjusted to 7 - 8. Serum and plasma samples can be assayed directly. Cell culture media should be diluted 5-10 fold in dH_2O prior to assay.

Colorimetric Procedure

1. Standards and Samples. Prepare a 1000 µM NH_3 Standard Premix by mixing 15 µL of the 20 mM Standard and 285 µL dH_2O. Dilute Standard as follows.

No	Premix + dH_2O	Vol (µL)	NH_3 (µM)
1	100 µL + 0 µL	100	1000
2	60 µL + 40 µL	100	600
3	30 µL + 70 µL	100	300
4	0 µL + 100 µL	100	0

   Transfer 20 µL standards into separate wells of a clear, flat-bottom 96-well plate.

   Transfer 20 µL of each sample into two separate wells, one serving as a sample blank well (RBLANK) and one as a sample well (R_{SAMPLE}).

2. Enzyme Reaction. For each standard and sample well, prepare Working Reagent by mixing 180 µL Assay Buffer, 1 µL Enzyme, 8 µL reconstituted NADH Reagent and 1 µL Ketoglutarate.
3. Add 180 µL of Working Reagent to the four Standards and the Sample Wells.
4. Prepare blank control reagent by mixing 180 µL Assay Buffer, 8 µL reconstituted NADH Reagent and 1 µL Ketoglutarate (No Enzyme).
5. Add 180 µL Blank control reagent only to the Sample Blank Wells. Tap plate to mix. Incubate 30 min at room temperature.
6. Read OD_{340nm}.

Fluorometric Procedure

The fluorometric procedure is the same as for the colorimetric assay, except that a black, flat-bottom 96-well plate is used. After incubation for 30 min at room temperature, read fluorescence intensity at λ_{ex} = 350-360 nm and λ_{em} = 450 nm.

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Calculations

Subtract the standard values from the blank value (#4) and plot the DOD or DF against standard concentrations. Determine the slope and calculate the NH_3 concentration of Sample,

$$[\text{Ammonia}] = \frac{R_{BLANK} - R_{SAMPLE}}{\text{Slope } (\mu M^{-1})} \times n \quad (\mu M)$$

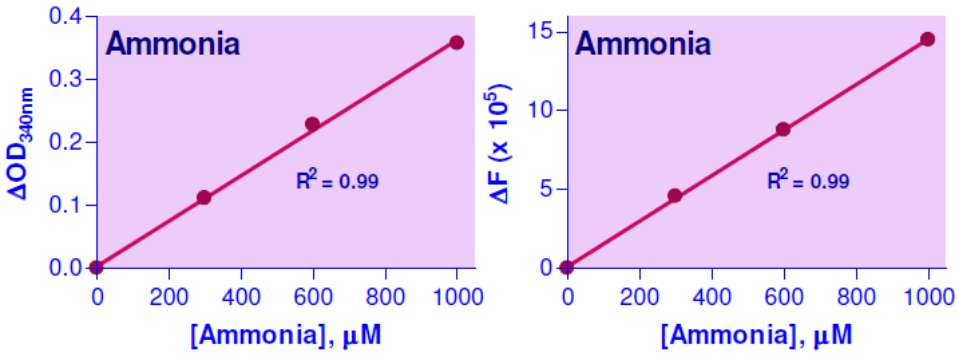
R_{SAMPLE} and R_{BLANK} are optical density or fluorescence intensity readings of the Sample and Sample Blank, respectively. *n* is the sample dilution factor.

Note: if the calculated NH3 concentration is higher than 1000 µM, dilute sample in dH2O and repeat assay. Multiply result by the dilution factor n.

Conversions: 1000 µM NH3 equals 1.7 mg/dL or 17 ppm.

Sample Data

Pipetting devices, and clear flat-bottom 96-well plates and optical density plate reader for colorimetric assays; black flat-bottom 96-well plate and fluorescence intensity plate reader for fluorometric assays.



96-well colorimetric assay *96-well fluorimetric assay*

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