

Glycerol Assay Kit (Colorimetric/Fluorometric)

LS-K214-200 (200 Tests) • Store at -20°C



Introduction

GLYCEROL [GLYCERIN or GLYCERINE, C₃H₅(OH)₃] is widely used in foods, beverages and pharmaceutical formulations. It is also a main byproduct of biodiesel production. Simple, direct and automation-ready procedures for measuring glycerol concentrations find wide applications. LSBio's glycerol assay uses a single Working Reagent that combine's glycerol kinase, glycerol phosphate oxidase and color reactions in one step. The color intensity of the reaction product at 570nm or fluorescence intensity at $\lambda_{em}/\lambda_{ex} = 585/530\text{nm}$ is directly proportional to glycerol concentration in the sample.

Key Features

- Sensitive and accurate. Use as little as 10 μL samples. Linear detection range in 96-well plate: 10 to 1000 μM (92 $\mu\text{g}/\text{dL}$ to 9.2 mg/dL) glycerol for colorimetric assays and 2 to 50 μM for fluorometric assays.
- Simple and convenient. The procedure involves addition of a single working reagent and incubation for 20 min at room temperature, compatible for HTS assays.
- Improved reagent stability. The optimized formulation has greatly enhanced the reagent and signal stability.

Applications

- Direct Assays: glycerol in biological samples (e.g. serum and plasma).
- Drug Discovery/Pharmacology: effects of drugs on glycerol metabolism.
- Food and Beverages: glycerol in food, beverages, pharmaceutical formulations etc.

Components

Component	K214-200
	200 Tests
Assay Buffer	24 mL
Enzyme Mix	500 μL
Dye Reagent	220 μL
ATP	250 μL
Standard (100 mM Glycerol)	100 μL

Materials Not Supplied

Pipetting devices, centrifuge tubes, clear flat-bottom 96-well plates, black 96-well plates (e.g. Corning Costar) and plate reader.

Storage

The kit is shipped on ice. Store all kit components at -20 °C.

FOR RESEARCH USE ONLY! Not for use in humans.

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Assay Procedure

Colorimetric 96-Well Procedure

Note: SH-group containing reagents (e.g. mercaptoethanol, DTT) may interfere with this assay and should be avoided in sample preparation.

1. Equilibrate all components to room temperature. Keep thawed Enzyme Mix in a refrigerator or on ice. Dilute standard in distilled water as follows (diluted standards can be used for future assays when stored refrigerated).

No	STD + H ₂ O	Vol (μL)	Glycerol (mM)
1	10 μL + 990 μL	1000	1.0
2	6 μL + 994 μL	1000	0.6
3	3 μL + 997 μL	1000	0.3
4	0 μL + 1000 μL	1000	0

Transfer 10 μL standards and 10 μL samples into separate wells of a clear 96-well plate.

2. For each reaction well, mix 100 μL Assay Buffer, 2 μL Enzyme Mix, 1 μL ATP and 1 μL Dye Reagent in a clean tube. This Working Reagent should be used on the same day of preparation. Transfer 100 μL Working Reagent into each reaction well. Tap plate to mix.
3. Incubate 20 min at room temperature. Read optical density at 570nm (550-585nm).

Note: if the Sample OD is higher than the Standard OD at 1.0 mM, dilute sample in water and repeat the assay. Multiply result by the dilution factor.

Fluorometric 96-Well Procedure

For fluorometric assays, the linear detection range is 2 to 50 μM glycerol. Mix 10 μL 100 mM Standard with 990 μL H₂O (final 1 mM).

No	1 mM STD + H ₂ O	Vol (μL)	Glycerol (mM)
1	50 μL + 950 μL	1000	0.050
2	30 μL + 970 μL	1000	0.030
3	15 μL + 985 μL	1000	0.015
4	0 μL + 1000 μL	1000	0

1. Dilute standards as above. Transfer 10 μL standards and 10 μL samples into separate wells of a black 96-well plate.
2. Add 100 μL Working Reagent (see Colorimetric Procedure). Tap plate to mix.
3. Incubate 20 min at room temperature and read fluorescence at $\lambda_{ex} = 530\text{nm}$ and $\lambda_{em} = 585\text{nm}$.
4. The glycerol concentration of Sample is calculated as

$$[\text{Glycerol}] = \frac{F_{\text{SAMPLE}} - F_{\text{H}_2\text{O}}}{\text{Slope}} \quad (\text{mM})$$

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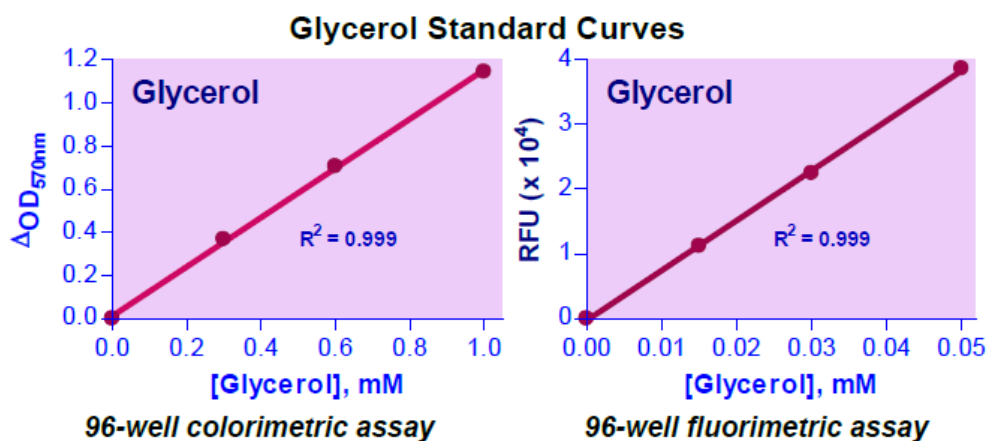
Calculations

Subtract blank OD (water, #4) from the standard OD values and plot the OD against standard concentrations. Determine the slope using linear regression fitting. The glycerol concentration of Sample is calculated as

$$[\text{Glycerol}] = \frac{F_{\text{SAMPLE}} - F_{\text{H}_2\text{O}}}{\text{Slope}} \quad (\text{mM})$$

OD_{SAMPLE} and $OD_{\text{H}_2\text{O}}$ are optical density values of the sample and water. Conversions: 1mM glycerol equals 9.2 mg/dL, 92 ppm.

Sample Data



Version: V.08.09.2018

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